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TECHNICAL REPORT DR-TR: 4-61

OF PA-101 AND PA-101D PRIMER COMPOSITIONS

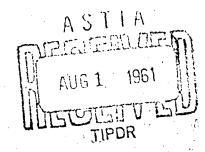
BY

WILLIAM F. MCGARRY

EDMUND P. VAIL

XEROX

COPY NO. 66 OF 100



JULY 1961

AMMUNITION GROUP
PICATINNY ARSENAL - DOVER, NEW JERSEY

TECHNICAL REPORT

EVALUATION OF STORAGE CHARACTERISTICS OF PA-101 AND PA-101D PRIMER COMPOSITIONS

ΒY

WILLIAM F, McGARRY
EDMUND P, VAIL

PROJECT NO:

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REPORT NO:

DR-TR: 4-61

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SECTION 1

INTRODUCTION

- 1. This work was conducted to determine the storage characteristics of PA-101 and PA-101D Primer Compositions.
- 2. PA-101 Primer Composition (53/22/10/5/10 basic lead styphnate/barium nitrate/antimony sulfide/tetracene/aluminum) was developed as a result of a request for a primer composition equal to or better than the one used to ignite the non-gaseous delay composition in the M42 Primer. PA-101 Primer Composition was found to have the characteristics that consistently initiate delay compositions in an obturated system--namely, low impulse and gas volume values (Reference 1). It was felt that the optimum loading pressure and storage characteristics of the composition should be determined before recommending it for standardization.
- 3. It was thought that the aluminum in the PA-101 Primer Composition might react with the oxidizers in the mixture during storage; therefore, a second primer composition designated PA-101D--was prepared using dichromated aluminum to prevent such possible reaction. The optimum consolidation pressure and charge weight for both the PA-101 and PA-101D Primer Compositions were established (Reference 2).
- 4. The work on PA-101 Primer Composition was to be continued as soon as funds were available. Equal quantities of M42 Primer metal parts were loaded with either Composition PA-101 or with the composition containing the dichromated aluminum. Each primer contained 20 milligrams of primer composition consolidated at 6,000 psi. Staircase sensitivity and gas volume-impulse tests were conducted at start of the program and at

regular intervals during the year while the primers were undergoing elevated and ambient temperature storage. The burning times of M204A2 Hand Grenade Fuzes assembled with the M42 Primers loaded with PA-101 and PA-101D Primer Compositions were determined before and after surveillance storage.

SECTION II

SUMMARY

- 1. Primer compositions PA-101 and PA-101D proved to have low impulse and gas volume values. These are two characteristics of a primer composition that consistently initiate delay compositions in an obturated system.
- 2. Equal quantities of M42 Primer metal parts were loaded with either Composition PA-101 or PA-101D and subjected to one year's storage at either ambient or elevated temperatures. Each primer contained 20 milligrams of primer composition consolidated at 6,000 psi. Staircase sensitivity and gas volume impulse tests were conducted at inception of the program and at regular intervals during the year, paralleling the approach with previously tested M42 Primers containing the standard composition.
- 3. Burning times of both primers (Composition PA-101 and PA-101D) assembled in M204A2 Hand Grenade Fuzes were determined before and after surveillance storage.

SECTION III

CONCLUSIONS

PA-101 and PA-101D Primer Compositions loaded in M42 Primer metal parts are stable when subjected to storage at extreme temperatures over a period as long as one year. M42 Primers when loaded with 20 milligrams of either PA-101 or PA-101D Primer Composition and consolidated at 6,000 psi will consistently initiate Type II zirconium-nickel alloy delay compositions. The PA-101D dichromated composition appears to be equivalent to plain PA-101 and in addition has better flow properaties.

SECTION IV

RECOMMENDATION

PA-101D Primer Composition should be used as the standard primer composition for initiating Type II Zirconium-nickel Alloy Delay Composition.

SECTION V

STUDY

1. The materials used in the work covered by this report include:

Basic lead styphnate, 90% min. through No. 325 mesh--U.S. Standard

Sieves--(commercial).

Barium nitrate, Class A, 99% min. through No. 70 mesh, PA-PD-253, 4/1/53.

Antimony sulfide, Class E, PA-PD-494, 1/15/54.

Tetracene, 50% min. through No. 325 mesh.

Aluminum, Type II, 200-325 nominal mesh size, MIL-P-14067, 5/16/55.

Composition, Delay, Type II, MIL-C-13739 (Ord), 11/2/54.

Lacquer, Egyptian, No. 4707, Commercial.

- 2. The PA-101 Primer Composition was prepared by blending 53% basic lead styphnate, 22% barium nitrate, 10% antimony sulfide, 5% tetracene and 10% aluminum. The PA-101D Primer Composition was prepared the same way except that the aluminum was dichromated before being blended with the other ingredients. The method used for dichromating the aluminum was essentially Dow Chemical Treatment No. 7 detailed on page 50 of Reference 3.
- 3. Equal quantities of M42 Primers containing PA-101 and PA-101D primer compositions were loaded by pressing 20 milligrams of dry primer composition into each primer cup at 6,000 psi. (The dichromated composition flowed more freely of the two compositions and therefore was easier to load). A drop of Egyptian Lacquer was placed on the charge followed by a foil paper disc of .005 inch thickness and then an anvil was pressed in place. Drawing 74-2-66 (Page C-1) was used as a guide. The primers were dried for 24 hours at a temperature of 55°C.

- 4. Two groups of 150 each of the M204A2 Hand Grenade Fuzes were assembled using M42 primers containing PA-101 Primer Composition in one group and primers containing PA-101D Primer Composition in the other group. The fuzes were assembled in accordance with Drawing F82-1-87 (Page C-2) except that in place of the detonator assembly (pc. mk 82-1-87E), the C-58 Non-electric Blasting Cup was used. Fifty fuzes from each of these groups were conditioned at -65°F, ambient temperature, and at 160°F for 16 hours before testing for burning times.
- 5. Each M204A2 Hand Grenade Fuze was clamped in the test fixture before functioning so that the delay charge column would not be damaged by dropping the fuze—the chief concern being the ability of the primer to initiate Type II Delay Composition—With the exception of clamping the fuze in the test fixture, the fuze functioning tests were conducted as outlined in paragraph 4.5 of Reference 6. The standard deviations in burning time for the PA-101 Composition were 0.10, 0.15, and 0.15 second at -65°F, ambient temperature, and 160°F respectively; those for the PA-101D Composition were 0.15, 0.13, and 0.12 second respectively. These tests were repeated using M42 Primers that had been stored for one year at 160°F and at ambient temperature. The detailed results of these tests are in Table I.
- 6. Frequency distribution graphs of the burning times recorded in these tests are in Figures 1-3. The same data is presented in Figure 4 to smaller scale, but on one page for easier comparison.
- 7. One hundred M42 Primers from each group were tested at the start of this program to determine the 50% functioning height and 50 M42 Primers from each group were used to determine the impulse and gas volume.

- 8. The staircase sensitivity test was performed by starting at an estimated 50% functioning height, using available sensitivity data as a guide. The step interval used was one inch. The results were evaluated in accordance with the procedure outlined in Reference 4.
- 9 The impulse and gas volume tests were run on the standard Picatinny Arsenal Impulse and Gas Volume Apparatus. The procedure followed was described in paragraph F7 of Reference 5 except that a value of 48 inch-ounces was used instead of the specified 56 inch-ounces. For both the sensitivity tests and the gas-volume impulse tests, the primers were pressed into the test adapters using a dead load of 200 pounds.
- 10. Primers loaded with the PA-101 Primer Composition produced an impulse and gas volume of 1.90 inches and 0.35 millimeters respectively-compared to 1.74 inches and 0.34 millimeters for similar primers using the PA-101D Primer Composition. Equal quantities of the remaining two groups of M42 Primers were placed in storage at 160°F and at ambient temperature
- 11. The sensitivity test and the impulse and gas volume tests were repeated with both groups of primers that were subjected to high temperature storage after one, two, four, six and twelve-month storage and with both groups after two, six and twelve-month storage at ambient temperature.

 The detailed test results are in Table 2.
- 12. Comparison of test results (Table 1 and 2) shows little difference in the characteristics of plain and dichromated PA-101 Primer Composition. Both compositions proved stable when exposed to extreme temperatures over a period of time. The slight difference seemed to be in favor of the dichromated composition. This composition was easier to load because it

flowed more freely than the plain composition. The weight of the charge and the consolidation pressure were determined in earlier work with these compositions. The M204A2 Hand Grenade Fuzes were clamped in the test device and functioned to prevent shock to the delay charge column and fuze metal parts which occurs when fuzes are dropped on steel plates. Type II Delay Composition was used since it is more difficult to ignite than the Type I Delay Composition which had been used in earlier tests of PA-101 primer composition.

REFERENCES

- 1. T. W. Stevens and K. G. Sheffield, <u>PA-101 Primer Mixture for Initiating Delay Compositions</u>, Explosives Development Section Report No. 10, April 1957.
- 2. K. G. Sheffield and W. F. McGarry, <u>PA-101 and PA-101-D Primer Mixtures</u>, Explosives Development Section Report No. 24, March 1958.
- 3. Magnesium Finishing, Magnesium Department, The Dow Chemical Company, Midland, Michigan. 1955.
- 4. Statistical Analysis for a New Procedure in Sensitivity Experiments, AMP Report No. 101.1R, July 1944.
- 5. Primer, Percussion, M29, Specification MIL-P-2496, 9 October 1950 with ANR 21266-S, 18 August 1953.
- 6. Fuzes, Grenade, Hand, M204A2, M206A2, and Practice, M205A2, Loading, Assembling and Packing, Specification OAC-PD-4 (Rev 3), 27 March 1957.

APPENDICES

APPENDIX A
DATA TABLES

APPENDIX A

DATA TABLES

APPENDIX A

DATA TABLES

TABLE 1

BURNING TIMES OF M204A2 HAND GRENADE FUZES

												•		200	0.00	> =		
			MIX BEIMER MIX	DPIN	N a	X		_				7	7-101-1	T.	PA-101-D PRIMER MAY	۲.		
			101-04	*****		ı					,		3	977	+	-	176 1400	ų
	,		126	12 @ AME T		12	12 @ 160 F	<u></u>	Storage Time, Months		2		271	16 @ AMB. 1.	-		2	
	>				-					100		4.071	46.5	ALLA	140.6	1.59	AMB	1.09
	1	100	3.091 AVA 3.57 3.071 2	AMB	160.5	-65°F	AMB	160 F	LESTE AMB 160 F Conditioning Temperatures -05 F AMB 180 F -05 F AMB 180 F	1,00-	AMP	1001	¥ 60-	3100		3	2	•
-65 F	AMB	1001	- 60							3	2	C,	0.5	50	20	50 50 50 50 50 50 50	20	20
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2	2	?	}	`				-		70	20	24.4	17	4	20	4 66	4 57	4 48
	7.	4 55	1 00 1 75 4 55 4 60 4 60 6.50	4 60			4.57	4.41	4, 70 4, 57 4, 41 Average Burning Time, sec 4, 00 4 10 4 55	00 -	?	5		3	:			
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	2	4 86	c 1 c c 20 4 84 4 84 4 95 4.70	4 95		4.89 4.81 4.62	4.81	4.62	Maximum, sec	30.0	r r		:	:	:			
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SENSITIVITY, IMPULSE AND GAS VOLUME TESTS OF M42 PRIMERS

Storage, Time, Months 0
Storage Temperatures
Staircase Test (c)
Number Tested 100 100
Sensitivity, 50% Height Inch 2.76 2.01
Standard Deviation, Inch 0.54 0.40
All Fire Level, Inch 4.38 3.21
No Fire Level, Inch 1.14 0.81
Impulse Test (d)
Number Tested 50
Average Inches 1.74 1.51
Standard Deviation, Inch 0.19 0.12
Maximum, Inch 2.20 2.20
Minimum, Inch 1.40 1.20
Gas Volume Test
Number Tested 50
Average, Milliliters 0.34 0.29
Standard Deviation, Milliliters 0.04 0.04
Maximum, Milliliters 0.40 0.35
Minimum, Milliliters 0.30 0.25

(a) 53% basic lead styphnate, 22% barium nitrate, 10% antimony sulfide, 5% tetracene, 10% aluminum.

⁽b) Same composition as PA101, but aluminum was dichromated prior to blending of the mix.

⁽c) 1.94-oz ball and 1-inch increments used.
(d) 8-oz ball dropped 6 inches.

APPENDIX B FREQUENCY DISTRIBUTION GRAPHS

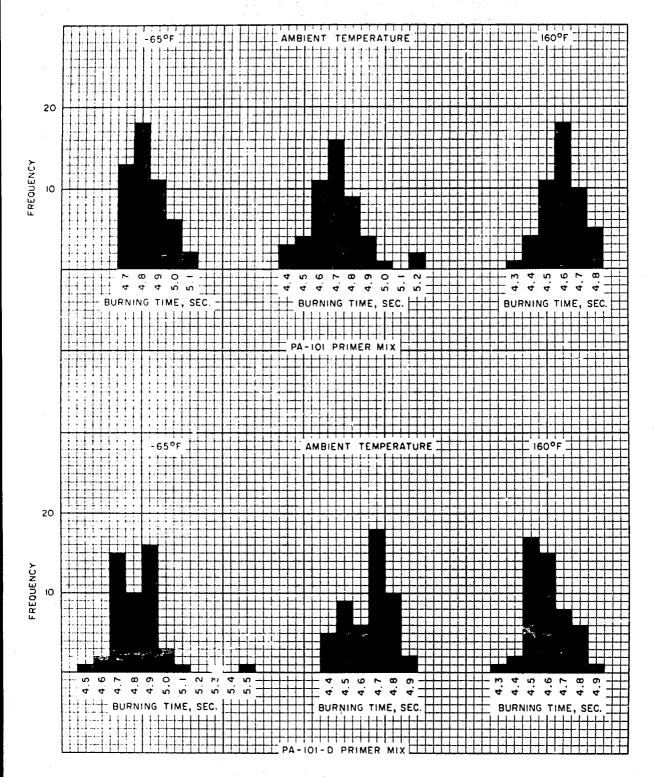


Figure 1. Frequency Distribution of Burning Time of M204A2 Hand Grenade Fuzes with Indicated Primer Mix. Primers Tested At Inception of Program

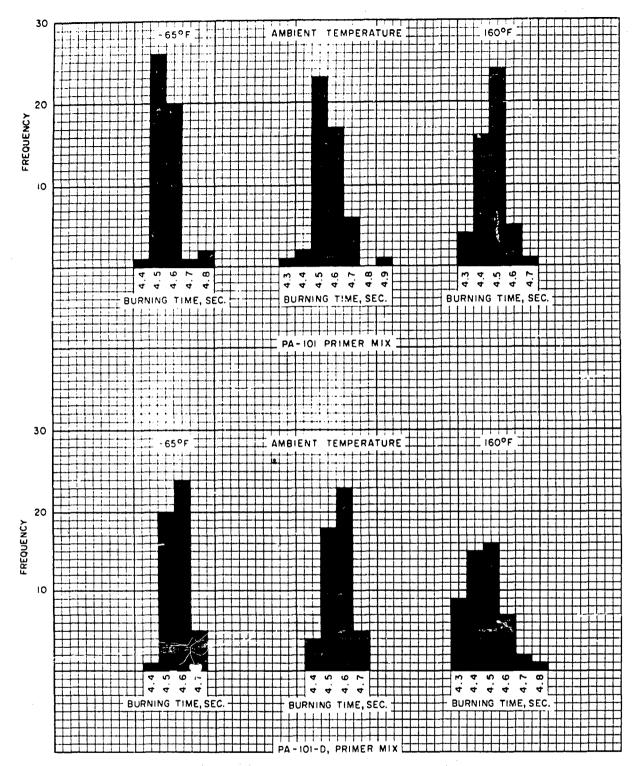


Figure 2. Frequency Distribution of Burning Time of M204A2 Hand Grenade Fuzes with Indicated Primer Mix. Primers Stored 12 Months At Ambient Temperature

Before Testing

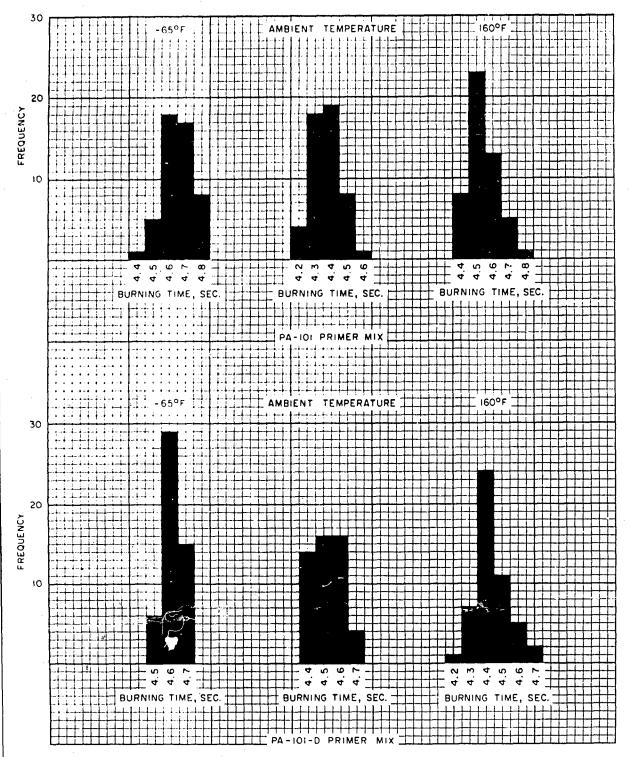


Figure 3. Frequency Distribution of Burning Time of M204A2 Hand Grenade Fuzes with Indicated Primer Mix. Primers Stored 12 Months at 160°F Before Testing

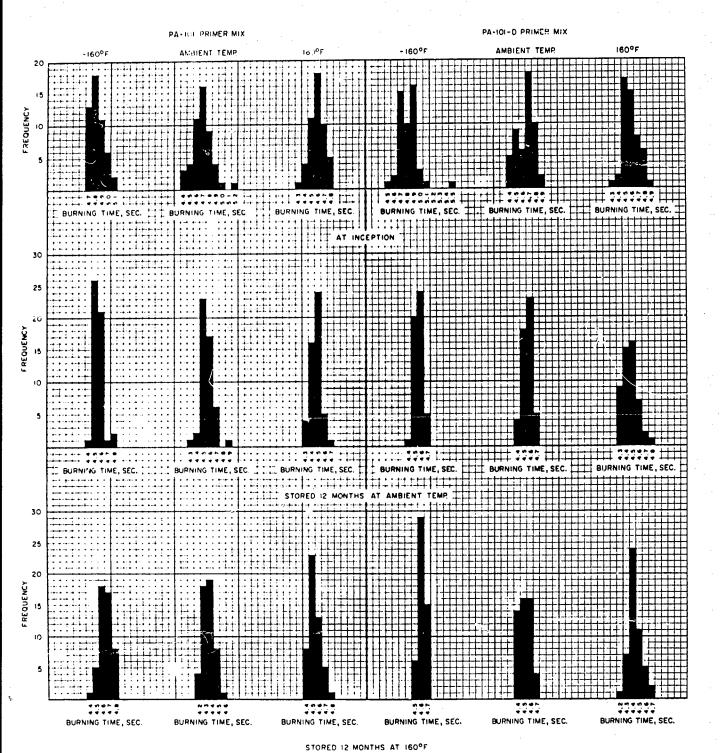
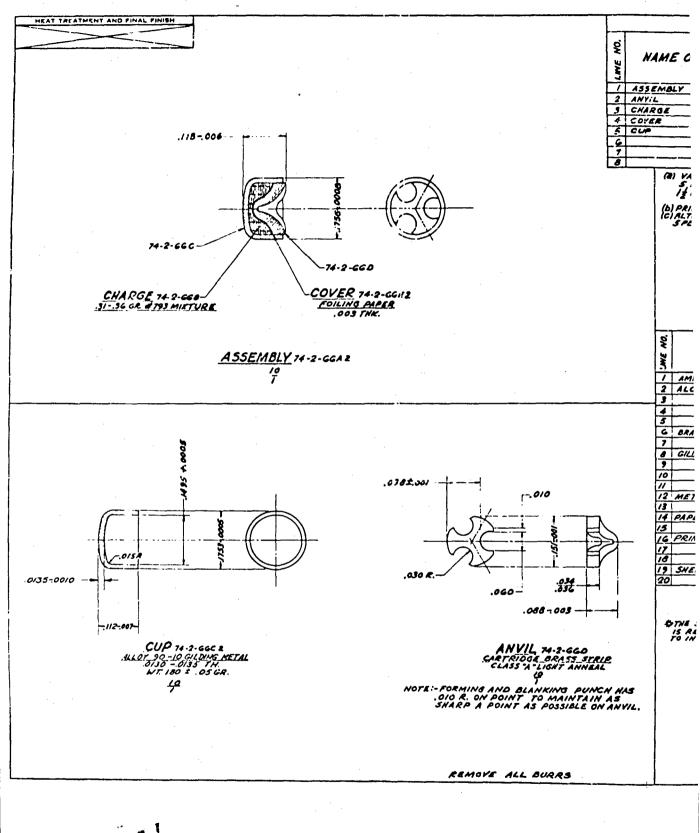


Figure 4. Frequency Distribution of Burning Time of M204A2 Hand Grenade Fuze with Indicated Primer Mix

APPENDIX C



_			LIST	07 7	AKIZ			,		,	PH
ς.		235	İ		MATERIAL	<u>'</u>		5 L3	5555	Ø	11
LINE NO.	NAME OF PART	NOW BE	PIECE- MARK	SIZE OR FORM	KIND	GRADE	SPEC. NUMBER	WEIGH	5730 CK A	REMAR	FL 2 RED
1	ASSEMBLY		74-2-66A				AX3-1288				}—
2	ANVIL	,	74-2-660	STRIP	CARTRIDGE BRASS	CLASS "A"	57-172-1	402.06			HOCK
3	CHARGE	/	74-2-668							(4)	ــــــــــــــــــــــــــــــــــــــ
4	COVER		74.2.66 €	SHEET	FOILING PAPER		50-11-24			(6)	J
5	cue		74-2-60 C		GILDING METAL		<u> </u>	1.982.46			.1
چ					ALCOHOL, ETHYL	2	4-1018		ļ	<u></u>	4
7					VARNISH, SHELLAG, SPECIAL	<u> </u>	<u> </u>	<u> </u>		(4)	_
8_	l						<u> </u>	<u></u>		<u></u>	_
	(B) VARMISH, SHEL S PINTS OF A If POUNDS OF C (b) PRIMER MIST WE (C) PRIMER MIST WE'S SPERCENT SHELL				4-1018 SPEC, TT-S-271 193 OR EQUAL MARKESMITE WITH						

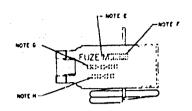
ME NO.	LIST OF SPECIFICATIONS	SPEC V:UI.1BER	REQUIRED BY		
1	AMMUNITION, EXCEPT SMALL ARMS AMM., GENERAL SPEC. FOR	50-0-1	ARS. 74-2-66		
2	ALCONOL, ETNYL	4-1018	DRG. 74-2-66		
,					
4					
5				*	
۵	BRASS, CARTRIOGE, SPEET AND STRIP	57-172-1	DR6.74-2-66		
7		ļ			
	GILD:NO METAL, SHEET, STRIP AND COIL	57-17/-1	DRG 74 2-66		RENADE, MANO, TZEI
<u>.</u>		 	ļ	AND FUZE, GRENA	ADE, HAND, TIB.
10	<u> </u>				
1/2		100 11 151			
13	METALS; GENERAL SPECIFICATIONS FOR INSPECTION OF	WW-M1-131	DRG. 74-2-66		
	PAPER FOLING	50-11-24	005.74-2-66		
15	PAPEA, FULLING	30-11-24			
_	PRIMER, M42	475-1288	000 74.866	M STANDAROVERD B	Y OCH 2679/ JAN, 18, 1945
17		7123-1600	D. 14-6-80	37777777777	
10			 	PPIM	ER M42
		l li	1		CR 1047
19		FF-5-27/	086.74-2-66		
	SHELLAC, ORANGE	FF-5-27/	DAS.74-2-GG		AND DETAILS
	SHELLAC, ORANGE	FF-5-27/	DA6.74-2-GG	ASSEMBLY MARCA 29,1944	
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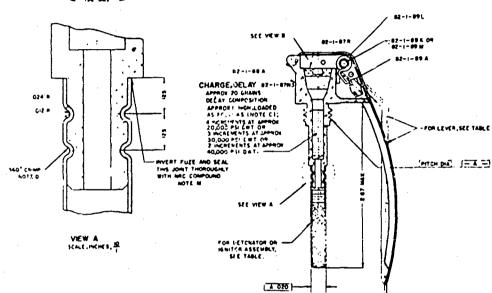




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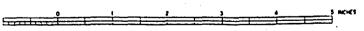
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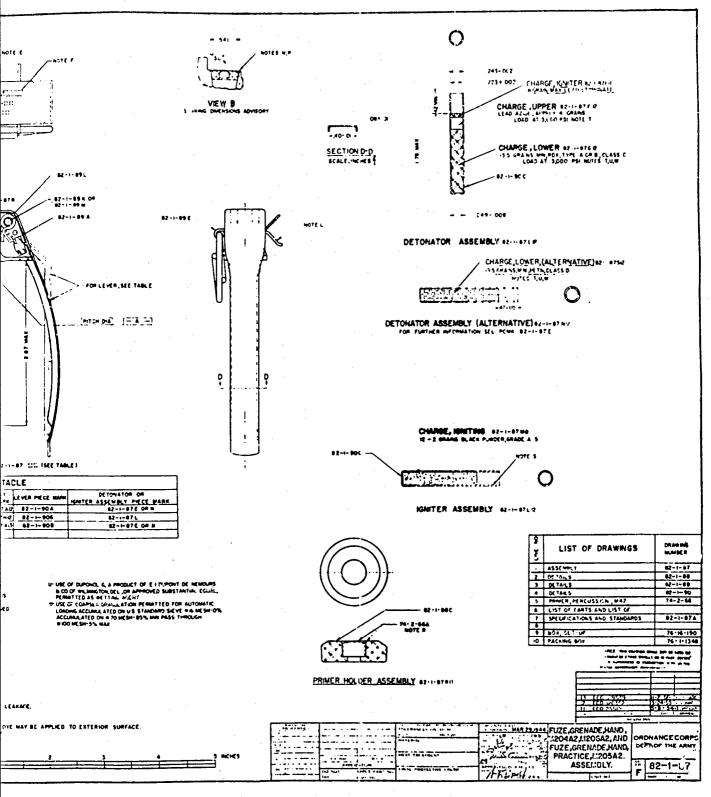


Figure 2

TABLE OF DISTRIBUTION

TABLE OF DISTRIBUTION

		Copy Number
1.	Commanding Officer	
	Picatinny Arsenal	
	Dover, New Jersey	1-5
	ATTN: ORDBB-DB1	
	ORDBB-DC2	6-10
	ORDBB-DR2	11-15
	ORDBB-DR4	16-30
	ORDBB~VS3	31 - 35
	ORDBB-DX1	36-37
2.	Commanding General	
	Ordnance Special Weapons-Ammunition Command	
	Dover, New Jersey	
	ATTN: J. E. Rainier	38 - 44
	T. J. Mahler	45-51
3.	Chief of Ordnance	
	Department of the Army	
	Washington 25, D. C.	
	ATTN: ORDIM	52-57
4,	Armed Services Technical Information Agency	
	Arlington Hall Station	
	Arlington 12, Virginia	58-68
		•
5.	Commanding Officer	
	Ordnance Ammunition Command	
	Joliet, Illinois	69-74
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6.	Commanding Officer	
	Frankford Arsenal	
	Bridge And Tacony Streets	
	Philadelphia 37, Pa.	75-80
7.	Commanding Officer	
	Diamond Ordnance Fuze Laboratories	
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9. Liaison Officer
British Joint Services
P. O. Box 680
Benjamin Franklin Station
Washington, D. C.

98-100

VIA:

Chief of Ordnance Department of the Army Washington 25, D. C. ATTN: ORDGU-SE ABSTRACT DATA

ABSTRACT

AD	Accession	No.

Picatinny Arsenal, Ammunition Group Dover, New Jersey

EVALUATION OF STORAGE CHARACTERISTICS OF PA-101 AND PA+101D, PRIMER COMPOSITIONS.

William F. McGarry and Edmund P. Vail

Technical Report DR-TR: 4-61, July 1961, 13 pp, drawings, appendices. Unclassified report

The two primer compositions proved to have low impulse and gas volume values. These characteristics make them suitable for use as initiating elements in obturated systems.

Surveillance tests demonstrated that the two compositions showed little difference, with the exception that the dichromated composition (PA-101D) proved easier to load because it flowed more freely than the plain composition.

UNCLASSIFIED

- 1. Primer compositions.
- 1. McGarry, William F.
- II. Vail, Edmund P.
- III. Project 252. 9501. 2600. 311.

UNITERMS

PA-101 PA -101D Primer Composition Lead styphnate Barium nitrate Antimony sulfide Tetracene Aluminum Delay Initiate Obturation Impulse Gas Volume Surveillance McGarry, William F. Vail, Edmund P. Project 252, 9501, 2600, 311

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